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Optimizing the catching of atoms or molecules in two-dimensional traps

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Single-photon cooling is a recently introduced method to cool atoms and molecules for which standard methods might not be applicable. We numerically examine this method in a two-dimensional wedge trap as well as in a two-dimensional harmonic trap. An element of the method is a small optical box with “diodic” walls which moves slowly through the external potential and catches atoms irreversibly. We show that the cooling efficiency of the method can be improved by optimizing the trajectory of this box.

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I. INTRODUCTION

Recently, a method has been introduced, called single-photon cooling [1,2], which allows one to cool atoms and molecules which cannot be handled in a standard way. The method is based on an atom diode or one-way barrier [3,4]. An atom diode is a device which can be passed by the atom only in one direction whereas the atom is reflected if coming from the opposite direction. Such a device has been studied theoretically [5–8] and also experimentally implemented as a realization of a Maxwell demon [9,10].

The idea of cooling based on an atom diode is explained in Refs. [1,7]. For illustration, we assume a simplified, idealized one-dimensional setting and an ensemble of noninteracting atoms which are moving classically in a harmonic potential trap (see Fig. 1). During the whole cooling process the atom diode is moved slowly through the trap from right to left with a constant velocity. In addition, we assume that the width of the diode is negligible. An atom can cross the atom diode only from left to right and after such a crossing the atom is trapped on the right-hand side of the diode.

For simplification let us look in the following at the process for a single atom of the ensemble (see Fig. 1 and also [7]). As the diode is moving much slower than the atom the diode captures every atom near its classical turning point (i.e., the atom has very low kinetic energy when it is caught). If the atom crosses the diode from left to right then this is an irreversible process, the atom cannot go back, that is, the atom is caught between the diode and the trap wall. Note that the implementations of an atom diode presented in [3,5–8] are designed in such a way that the internal state of the atom is changed during the passing of the diode but it will be restored at the end in such a way that the atom is in the same internal state before and after crossing the diode. The diode, which is in principle a semipenetrable barrier, behaves like a wall for the captured atom and continues moving to the left without changing its velocity. During this the atom is bouncing off the slowly moving diode and in such a way the kinetic energy of the atom is reduced. Note that the total energy is not conserved during this process (a Hamiltonian describing the system would be time dependent). So the diode transports the caught atom to the bottom of the potential without increasing the kinetic energy of the atom; on the contrary, the kinetic energy of the atom might be even more decreased. This can be heuristically compared with a ball bouncing on a horizontal racket where the racket is slowly moved down in the gravity field: Then the ball is not bouncing “more” if the racket is arrived at the bottom. Note that even if the atom has reached the bottom of the trap, diode still continues moving to the left. Finally, the atom is in the same external potential (this is different, for example, from [11]) and in the same internal state as it was initially, but its total energy has been reduced.

If the process acts on the whole ensemble of noninteracting atoms—of a given temperature initially—then the energy of every atom is reduced and therefore the ensemble is cooled.

It is important to underline that—because of the irreversible step—this cooling method is fundamentally different from velocity reduction by collision with a moving wall which is not “real” cooling [12].

Several variations of this cooling method have been proposed and applied in various experiments, both for cooling atoms [13–17] and for cooling molecules [18,19].

As a variant to the scheme shown in Fig. 1, a small optical dipole trap has been used in experiment [14]. The small trap is constructed in the form of a square box [14] and therefore we will simply call it “box” in this paper (not to be confused with the external potential trap). The key idea is that this small box is moved slowly through the external potential. In an ideal setting, the walls of the box consist of atom diodes: If the atom has crossed a wall of the box and the kinetic energy of the atom is lower than the threshold energy of the “diodic” wall (called the threshold energy of the box in the following) then the atom is irreversibly caught by the box. Note that in the ideal case the caught atom is in the same internal state as before the catching (see also [5–8]). In such a way, a similar effect as shown in Fig. 1 may be achieved.

It is important that the box is moved in such a way that it traps a maximal number of atoms. While this optimal trajectory of the box is straightforward in a one-dimensional setting, this is not obvious in higher dimensions. In [20], a two-dimensional wedge trap has been examined. In that paper the box was assumed to be at rest and the optimal position of the box was determined. The main goal of our paper is to extend this work [20] and show that by a slowly moving box the cooling efficiency can be improved compared to a
of a single atom in this trap is 

$$\omega$$

where $\omega$ is the frequency of the harmonic trap.

The second trap is a two-dimensional harmonic trap [see Fig. 2(b)]. The Hamilton function of a single atom in this trap is

$$H_{H} = \frac{1}{2m}(p_{x}^{2} + p_{y}^{2}) + \frac{m}{2}\omega^{2}(x^{2} + y^{2}),$$

where $\omega$ is the frequency of the harmonic trap.

In both cases, we approximate the motion of the atoms as a classical motion and in addition, we assume that the atoms inside the trap do not interact with each other. So the trajectory of a single atom can be calculated by solving Hamilton’s equations,

$$\frac{\partial x}{\partial t} = \frac{\partial H}{\partial p_{x}}, \frac{\partial y}{\partial t} = \frac{\partial H}{\partial p_{y}}, \frac{\partial p_{x}}{\partial t} = -\frac{\partial H}{\partial x}, \frac{\partial p_{y}}{\partial t} = -\frac{\partial H}{\partial y}.$$  

The initial state of a single atom should be distributed concerning the canonical probability distribution with temperature $T_i$

$$\rho_{i}(x,y,p_{x},p_{y}) = \frac{1}{Z} \exp \left[-\beta H(x,y,p_{x},p_{y})\right],$$

where $Z$ is the canonical partition function, $\beta = 1/(k_B T_i)$, and $k_B$ is the Boltzmann constant.

Our simplified model of the catching process is similar to and motivated by [20]. We model the “box,” as a square box in space with a width $2w_B$ in $x$ and in $y$ direction, that is, $w_B$ is the half width of the box. The box can move freely throughout the trap. We assume that the “diodic” walls of the box are infinitely thin. Moreover, the “diodic” walls of the box have a certain threshold energy $E_B$ such that their “diodic” behavior breaks down if the kinetic energy of the atom is above this threshold energy. If the atom has crossed—at some time—the walls of the box and the kinetic energy of the atom in the box rest frame is smaller than the threshold energy $E_B$ then the atom is caught irreversibly by the box for all times. If the kinetic energy of the atom is too large (larger than $E_B$) then the “diodic” property of walls breaks down and the atom can escape the box such that it might have a second chance later to be caught.

The algorithm is now the following: For a single numerical run, we choose the initial state of a single atom randomly with respect to the probability distribution Eq. (4). Then we calculate numerically the evolution of the atom until a final time $t_f$ while always checking if the atom is in the region of the box during its motion. If this is the case, we are checking if the kinetic energy of the atom in the box rest frame is smaller than the box threshold energy $E_B$. If this is true, then the atom is caught and the run is finished, otherwise the motion of the atom continues and it may get trapped later. We repeat such a single run $N_i = 10^6$ times, get the total number of trapped atoms $N_B$ and finally the relative number of trapped atoms (i.e., the fraction $F = N_B/N_i$).

The goal of this paper is to optimize this fraction $F$ by varying the trajectory of the box in the trap. As we will see now, from this follows also an optimization of the cooling efficiency if the initial conditions and the box parameters are fixed.

One can define a cooling efficiency as the compression of the phase-space density. Phase-space density in the context of cooling can be defined as the number of atoms in a box with sides of one “thermal de Broglie wavelength” [21]. The “thermal de Broglie wavelength” $\lambda$ can be defined as

$$\lambda = \frac{\hbar \sqrt{2\pi}}{\sqrt{m k_B T}},$$

where $m$ is the mass of the atom. The phase-space density in a two-dimensional setting is then defined as $\mu = n \lambda^2$, where $n$ is the spatial density of atoms. The relative change in phase-space density is therefore

$$\frac{\mu_f}{\mu_i} = \frac{n_f}{n_i} \frac{T_i}{T_f},$$

where $\mu_i$ is the initial and $\mu_f$ is the final phase-space density. $T_i$ and $T_f$ are the initial and final temperature, respectively.
The initial distribution will be a canonical one with temperature $T_i$. The initial spatial density $n_i$ is strictly speaking not constant in space. Nevertheless, we set approximately $n_i = N_i/A_i$, where $N_i$ is the initial number of atoms. To get $A_i$, we fix a small $\varepsilon > 0$ and define the initial region $A_i$ such that the atom is initially in $A_i$ with probability $1 - \varepsilon$. The final spatial density is set as $n_f = N_f/A_f$, where $N_f$ is the number of trapped atoms and $A_f = 4w_f^2$ is the area of the box.

The cooling efficiency $\eta$ is then defined as

$$\eta = \log_{10} \left( \frac{\mu_f}{\mu_i} \right) = \log_{10} \left( \frac{N_f/A_f}{N_i/A_i} \right).$$

Here $E_i = k_BT_i$. For simplicity, the final temperature is approximated by the threshold energy of the box (i.e., $T_f = E_B/k_B$). The cooling efficiency depends on the fraction of trapped atoms $F = N_f/N_i$, the area of the ratio of the thermal system to the area of the box, $A_i/A_f$, and the ratio of the initial to final energy, $E_i/E_B$:

$$\eta = \log_{10} \left( F \frac{A_f}{A_i} \frac{E_i}{E_B} \right).$$

If the initial conditions and the box parameters are fixed, it will be therefore sufficient to optimize the fraction of trapped atoms $F$. The goal of the rest of the paper will be to optimize this fraction by choosing different box trajectories.

III. BOX TRAJECTORIES IN A WEDGE TRAP

First we study the wedge trap [see Fig. 2(a)]. According to Eq. (4), the initial canonical distribution is given by

$$\rho_{i,w}(x,y,p_x,p_y) = \frac{m g^2 \beta^3}{4\pi \tan \alpha} \chi_f(x) \exp \left[ -\beta \left( \frac{p_x^2 + p_y^2}{2m} + mg y \right) \right],$$

where $\chi_f(x)$ is the indicator function with $\chi_f(x) = 1$ if $x \in J$ and $\chi_f(x) = 0$ if $x \notin J$. It is convenient to define a characteristic length $l$, a characteristic velocity $v$, and a characteristic time $\tau$ by

$$l = \frac{k_BT_i}{mg}, \quad v = \sqrt{\frac{k_BT_i}{m}}, \quad \tau = \sqrt{\frac{k_BT_i}{mg^2}}.$$

In the rest of the paper we are assuming $^{87}\text{Rb}$ atoms [i.e., $m = \text{mass}(^{87}\text{Rb}) = 1.44316 \times 10^{-25}$ kg]. The initial temperature should be $T_i = 100\mu K$ and $g = 9.78$ m/s$^2$ (equator). The characteristic values are then

$$l = 978\mu m, \quad v = 9.78\text{ cm/s}, \quad \tau = 10\text{ ms}.$$  

A. Box at rest

The center of the box is moving with a trajectory $[x_B(t), y_B(t)]$. As a reference case we first assume that the box center is placed at $x_B = 0$ at rest, that is, its velocity is $v_{B,x} = v_{B,y} = 0$. Figure 3(a) shows the trapping fraction $F$ versus different box positions $y_B$ for different combinations of wedge angle $\alpha$ and box half width $w_B$.

The optimal height $y_{op}$ which gives the maximal fraction of trapped atoms for different box half widths $w_B$ is shown in

FIG. 3. (Color online) Wedge trap with the box at rest at $x_B = 0$. (a) Fraction $F$ of trapped atoms versus box positions $y_B$: $\alpha = 30^\circ, w_B/l = 0.1$ (solid blue line); $\alpha = 45^\circ, w_B/l = 0.1$ (dashed red line); $\alpha = 60^\circ, w_B/l = 0.1$ (dotted green line); $\alpha = 30^\circ, w_B/l = 0.35$ (thick solid blue line); $\alpha = 45^\circ, w_B/l = 0.35$ (thick dashed red line); $\alpha = 60^\circ, w_B/l = 0.35$ (thick dotted green line). (b) Optimal position $y_{op}$ of the box at rest for different wedge angles (blue, red, respectively, green plus signs connected with lines), the black dots correspond to the straight line $y_{op} = w_B/(1 + \tan \alpha)/\tan \alpha$; see text for more details. $t_f/\tau = 20, E_B/E_i = 0.1$.

B. Box with linear motion

We want to examine if a moving box can produce a higher fraction of trapped atoms than the box at rest. First, we consider a linear motion of the box given by

$$x_B(t) = v_{B,x}(t - t_f/2), \quad y_B(t) = v_{B,y}(t - t_f/2) + y_{op}.$$

Here $y_{op}$ is chosen to be equal to the optimal value for the box at rest shown in Fig. 3(b) (for the corresponding wedge angle and box half width). Because of mirror symmetry $(x \leftrightarrow -x)$ of the setting we can restrict to the case $x_{B,x} > 0$.

Figure 4 shows the resulting fraction $F$ versus different box velocities for different wedge angles. The box half width is fixed at $w_B/l = 0.35$. The square symbols mark the maximal fraction $F$ which can be achieved with a box at rest while the dots mark the maximal fraction which can be achieved with a box moving with the trajectory Eq. (12). It is clearly seen that the linear moving box can capture a larger fraction $F$ of atoms than the box at rest (for all the three different wedge angles $\alpha = 30^\circ, 45^\circ$, and $60^\circ$).

The optimal velocity in the first case $\alpha = 30^\circ$ [Fig. 4(a)] is

$$v_{B,x}/v = 0.06 \approx 0.13\sin 30^\circ, \quad v_{B,y}/v = 0.12 \approx 0.13\cos 30^\circ.$$  

In the case $\alpha = 45^\circ$ [Fig. 4(b)], we get for the optimal velocity,

$$v_{B,x}/v = 0.08 \approx 0.11\sin 45^\circ, \quad v_{B,y}/v = 0.08 \approx 0.11\cos 45^\circ.$$
FIG. 4. Wedge trap. Box moving with the linear trajectory Eq. (12). Fraction $F$ of trapped atoms versus box velocities: (a) wedge angle $\alpha = 30^\circ$, $y_{op}/l = 0.7$; (b) wedge angle $\alpha = 45^\circ$, $y_{op}/l = 0.6$; (c) wedge angle $\alpha = 60^\circ$, $y_{op}/l = 0.75$, $t_f/\tau = 20$, $E_B/E_i = 0.1$, $w_B/l = 0.35$. The square symbols mark the maximal fraction for a box at rest; the dots mark the maximal fraction for a linear moving box.

It is important to notice from Eqs. (13) and (14) that we get in both cases the maximal fraction if the box is approximately moving parallel to one wedge side. This is different in the case $\alpha = 60^\circ$ [Fig. 4(c)]. The optimal velocities are now

$$v_{B,x}/v = 0.04 \approx 0.072 \sin 34^\circ,$$
$$v_{B,y}/v = 0.06 \approx 0.072 \cos 34^\circ,$$

that is, the box motion is not parallel to the wedge side in this case. Nevertheless, the linear moving box traps in all cases more atoms than the corresponding box at rest.

Therefore, the box moving with the linear trajectory enhances the trapping fraction than the box at rest. In the following, we will study the linear moving box in more detail to see if this result remains true for other box half widths.

We first look at the cases of a wedge angle $\alpha = 30^\circ$ and $\alpha = 45^\circ$, respectively. Figures 6 and 7 show the fraction $F$ of trapped atoms versus different box half widths $w_B$. Figures 6(a) and 7(a) correspond to the final time $t_f/\tau = 20$ while Figs. 6(b) and 7(b) correspond to $t_f/\tau = 40$.

The results for a box at rest ($x_B = 0, v_{B,x} = v_{B,y} = 0$) are shown as a reference case for $t_f/\tau = 20$ as well as for $t_f/\tau = 40$ (plus signs connected by thick blue line) with the box coordinate $y_B = y_{op}$ and $y_{op}$ shown in Fig. 3(b) (which is optimal for $t_f/\tau = 20$). It can be seen in Figs. 6 and 7 that for a box at rest the fraction does not change significantly if the final time $t_f$ increases.

Motivated by the previous results, we are again consider a box which moves linearly in the direction of the right wedge side (i.e., its velocity is $v_{B,x} = v \sin \alpha, v_{B,y} = v \cos \alpha$). The box trajectory should cross the $y$ axis at $y_{op}$, where $y_{op}$ is again the position shown in Fig. 3(b). Nevertheless, the box should now start at time $t = 0$ directly outside the wedge trap; see Fig. 5. Note that therefore the $y$ axis is no longer crossed at time $t_f/2$ by the box center. The resulting trajectory is

$$x_B(t) = v \sin \alpha t - \frac{1}{2} [w_B + (w_B + y_{op}) \tan \alpha],$$
$$y_B(t) = v \cos \alpha t - \frac{1}{\tau} (w_B - y_{op} + w_B \cot \alpha).$$

FIG. 6. (Color online) Wedge trap, $\alpha = 30^\circ$. Fraction $F$ of trapped atoms versus box half widths $w_B$: (a) $t_f/\tau = 20$, (b) $t_f/\tau = 40$. Different box trajectories are as follows: box at rest (plus signs connected by thick blue line), box moving with trajectory Eq. (16) (crosses connected by red dashed line), box moving with analytical trajectory Eq. (17) (dots connected by thick green dotted line), box moving with wriggle trajectory Eq. (18) (circles connected by black solid line), $v/v = 0.13$, $y_{wo}/l = 2.0$, $aw/\tau = 0.25$, $E_B/E_i = 0.1$. 

FIG. 5. (Color online) Schematic representation of the box trajectories Eqs. (16) and (17); the initial position of the box is shown (see text for further details).
The value of \( y_{\text{op}} \) can be obtained by numerical optimization. However, these box trajectories still depend on the values \( y_{\text{op}} \) which have to be found by numerical optimization. The goal is now to find an analytical approximation for \( y_{\text{op}} \).

The results for such a moving box for \( t_f/\tau = 20 \) as well as for \( t_f/\tau = 40 \) are shown in Figs. 6 and 7 (crosses connected by red dashed line). It can be seen that the fraction \( F \) is—in all cases—much larger for such a linearly moving box compared to the box at rest.

However, these box trajectories still depend on the values \( y_{\text{op}} \) which have to be obtained by numerical optimization. The goal is now to find an analytical approximation for \( y_{\text{op}} \). Therefore, we are considering a box moving parallel to the right wedge side while touching the wedge side with its right lower corner (see also Fig. 5). Then we get the analytical value \( y_{\text{op}} = y_B(1 + \tan \alpha)/\tan \alpha \). Note that these values for \( y_{\text{op}} \) are also plotted in Fig. 3(b) and we can see that this is also a rough approximation for the numerical determined optimal position \( y_{\text{op}} \). This “analytical” box trajectory is now

\[
\begin{align*}
    x_B(t) &= v \sin \alpha t - w_B(1 + \tan \alpha), \\
    y_B(t) &= v \cos \alpha t.
\end{align*}
\]

The resulting fraction \( F \) using the box trajectory Eq. (17) can be seen in Figs. 6 and 7 (dots connected by thick green dotted line). We find that the box traps even more atoms than with the trajectory Eq. (16) considered in the previous paragraph and it has the advantage that no numerical determined value of \( y_{\text{op}} \) is required.

For completeness, the fraction \( F \) versus the box half widths for \( \alpha = 60^\circ \) is shown in Fig. 8. The results are shown for a box at rest (plus signs connected by a thick blue line) and a box moving linearly with trajectory Eq. (16) as well as Eq. (17). The value of \( y_{\text{op}} \) in Eq. (16) is chosen from Fig. 3(b). We see that the moving box catches more atoms than the box at rest. In contrast to the cases \( \alpha = 30^\circ \) and \( \alpha = 45^\circ \), the analytical box trajectory Eq. (17) is here a good choice only for \( t_f/\tau = 40 \).

C. Box moving with wriggled trajectory

We want to underline that a linear trajectory might not be optimal. However, the advantage of a linear motion of the box is that it might be easier to implement experimentally a linear motion than a more complicated motion of the box.

Nevertheless, as an example, we shall also try a different trajectory of the box, a wriggled one, given by

\[
\begin{align*}
    x_B(t) &= y_B(t) \tan \alpha \cos(\omega_W t), \\
    y_B(t) &= y_W + [w_B - y_W] t / t_f.
\end{align*}
\]

The box center starts at the right wall of the trap (i.e., \( x_B(0) = y_W \tan \alpha, y_B(0) = y_W \)) and then moves down wriggling and ends at \( y(t_f) = y_B \) (i.e., the box half width). The resulting trapping fractions \( F \) can also be seen in Figs. 6–8 (circles connected by black solid line). The parameters for the box trajectory have been chosen such that the velocity of the box is limited \( v/\nu < 0.265 \) at all times. In the cases \( \alpha = 30^\circ \) and \( \alpha = 45^\circ \) we get an increased fraction of trapped atoms for \( t_f/\tau = 40 \).

IV. OPTIMIZING BOX TRAJECTORIES FOR A HARMONIC TRAP

Now, we study the harmonic trap with frequencies \( \omega_x = \omega_y = \omega \) [see Fig. 2(b)]. The initial state of an atom is chosen concerning the canonical distribution,

\[
\rho_{\text{H}}(x,y,p_x,p_y) = \frac{\omega_x^2 \beta^2}{4\pi^2} \exp \left\{ -\beta \left[ (p_x^2 + p_y^2)/(2m) + m \omega^2 (x^2 + y^2)/2 \right] \right\}.
\]
Again, it is convenient to define a characteristic length $l$, a characteristic velocity $v$, and a characteristic time $\tau$ by

$$l = \frac{1}{\omega} \sqrt{\frac{k_B T_i}{m}}, \quad v = \sqrt{\frac{k_B T_i}{m}}, \quad \tau = \frac{1}{\omega}. \quad (20)$$

In the rest of the paper we are again assuming $^{87}$Rb atoms, the initial temperature shall be $T_i = 100 \mu K$ and $\omega = 50 \times 2\pi$ Hz. The characteristic parameters are then

$$l = 311 \mu m, \quad v = 9.78 \text{ cm/s}, \quad \tau = 3.18 \text{ ms}. \quad (21)$$

A. Box at rest

As a reference, we first look again at the case that the box center is placed at $x_B = 0$ at rest (i.e., $v_{B,x} = v_{B,y} = 0$). The box threshold energy is $E_B/E_i = 0.1$ and the final time is $t_f/\tau = 60$. The box coordinate $y_B = y_{op}$ is numerically optimized such that the resulting fraction of trapped atoms $F$ is maximal. Figure 9(a) shows the trapping fraction $F$ for different box half widths $w_B$ (plus signs connected by thick blue line). The optimized box coordinate $y_{op}$ is shown in Fig. 9(b) (plus signs connected by thick blue line). The error bars are defined by the range in which the maximal trapping fraction $F$ decays by an amount of $1/\sqrt{N}$, where $N$ is the number of particles used in the numerical simulation. From the error bars, we can see that for small $w_B$ the result is relatively independent of the exact box position $y_B$.

B. Box with linear motion

We want to examine if a linearly moving box can produce a higher fraction $F$ of trapped atoms in the case of a harmonic trap.

In Fig. 10, the resulting fraction $F$ is shown for the box half width $w_B/l = 0.2$. The box symbol marks the maximal fraction for a box at rest while the dot marks the maximal fraction for a linearly moving box. It can be seen that the linearly moving box traps more atoms than the box at rest.

We also calculated the fraction for different box half widths; the result can be seen in Fig. 9(a). For every box half width, the position $y_c$ and the velocity $v_{B,y}$ have been optimized and the optimal values are shown in Figs. 9(b) and 9(c) (crosses connected by red dashed line). The error bars are defined in the same way as explained in the last subsection. In Fig. 9(a) it can be seen that in all cases the moving box traps significantly more atoms than the box at rest.

For the case that the box is moving in such a way that its trajectory crosses the point $x = 0$, $y = y_c$ at time $t_f/2$. Because of symmetry, we can restrict to the case $v_{B,x} > 0, v_{B,y} = 0$ (if we neglect the rotation of the box itself). We consider a linear trajectory of the moving box as

$$x_B(t) = v_{B,x}(t - t_f/2), \quad y_B(t) = y_c. \quad (22)$$

The resulting fraction $F$ for a box moving with trajectory Eq. (22) is also shown in Fig. 9(a) (dots connected by thick green dotted line on top of the red line); it is indistinguishable from the result obtained with trajectory Eq. (22) with optimized parameters $y_c$ and $v_{B,x}$.

In the following we will show that this trajectory Eq. (25) is also a very good choice for different box threshold energies.
Just as an example, we try a different, more complicated box trajectory. The box is moving with a helix trajectory given by

\[ x_B(t) = x_H \left( 1 - t/t_f \right) \cos(\omega_H t), \]
\[ y_B(t) = x_H \left( 1 - t/t_f \right) \sin(\omega_H t), \]

where \( x_H \) and \( \omega_H \) are the key parameters of the box trajectory. The results can be seen in Fig. 9 as well as in Fig. 11 (circles connected by black solid line). The parameters of the helix trajectory are \( x_H/l = 1.9 \) and \( \omega_H = 0.1 \). For larger box half width \( w_B \) [see Fig. 11(b)] this trajectory helps to get a larger fraction of trapped atoms while there is no improvement for smaller box half width.

V. SUMMARY AND OUTLOOK

We have examined a model for cooling in a two-dimensional wedge trap and a two-dimensional harmonic trap. During the cooling procedure, the atoms are captured by a small area surrounded by “diodic” walls, called “box,” which moves through the wedge trap and the harmonic trap, respectively. We have examined different box trajectories with the goal to maximize the fraction of trapped atoms, this leads also to an increased cooling efficiency. We have shown that the fraction of caught atoms can be increased using a moving box compared to a box at rest which was examined in an earlier work [20]. We have also optimized the parameters of the box trajectory where we restricted ourselves mainly to linear box motions due to its possible easier experimental implementation. In a future work, we will consider the question of the optimal general box trajectory in more detail and also taking quantum effects into account.

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